

thomas c. memahon, director

### BUZZARDS BAY

### 1978 - 1979

### WASTEWATER DISCHARGE SURVEY DATA

### Prepared by

Water Quality and Research Section Massachusetts Division of Water Pollution Control

> Westborough, Massachusetts November 1979

PUBLICATION #11,676-33-50-12-79-CR

Approved by: Alfred C. Holland Purchasing Agent

## TABLE OF CONTENTS

ITEM	PAGE
Introduction	2
List of Wastewater Discharges	3
Municipal Wastewater Discharges	
Dartmouth	6
Fairhaven	8
Marion	10
New Bedford	12
Wareham	14
Industrial and Institutional Wastewater Discharges	
Acushnet Company, Golf Division	16
Acushnet Company, Rubber Division	18
Chamberland Manufacturing Company	20
Lincoln Park Amusement Company	22
Massachusetts Maritime Academy	24
Old Rochester High School	26
Revere Copper and Brass	28
Glossary	30

#### INTRODUCTION

This report presents samples collected from the major municipal wastewater discharges and some industrial wastewater discharges. The sampling occurred from October 23 to October 26, 1978, and again from August 27 to August 29, 1979. Sampling was conducted by personnel of the Water Quality and Research Section of the Division of Water Pollution Control in Westborough, with assistance from the personnel at the municipal and industrial wastewater treatment facilities.

Composite samples were collected at the municipal wastewater treatment plants. Grab samples were taken from the industrial plants and the minor sewage treatment facilities. Flow rates were obtained from plant records.

All samples were conveyed to the Lawrence Experiment Station of the Massachusetts Department of Environmental Quality Engineering for analysis. Samples collected were analyzed in accordance with <u>APHA's</u> <u>Standard Methods for Examination of Water and Wastewater</u>, current edition.

2

## BUZZARDS BAY BASIN WASTEWATER DISCHARGES

### NUMBER

### MUNICIPAL

- Dartmouth Wastewater Treatment Plant
  Fairhaven Wastewater Treatment Plant
  Marion Wastewater Treatment Plant
- 4 New Bedford Wastewater Treatment Plant
- 5 Wareham Wastewater Treatment Plant

### INDUSTRIAL, BUSINESS, INSTITUTIONAL WASTEWATER DISCHARGES

6	Acushnet Company, Golf Division, Acushnet
7	Acushnet Company, Rubber Division, New Bedford
8	Chamberland Manufacturing Company, New Bedford
9	Lincoln Park Amusement Company, Dartmouth
10	Massachusetts Maritime Academy, Bourne
11	Old Rochester High School, Mattapoisett
12	Revere Copper & Brass, New Bedford



FIGURE I



## DARTMOUTH WASTEWATER TREATMENT PLANT

LOCATION:	759 Russells Mill Road South Dartmouth, Massachusetts
RECEIVING WATER:	Buzzards Bay
CAPACITY:	2.1 MGD
TREATMENT PROCESS:	Extended aeration
	Chlorination
SLUDGE DISPOSAL:	Landfilled at municipal site
TYPE OF SAMPLE:	Two 24-hour composites of the effluent
	Two grab samples after chlorination for coliform bacteria
	One 24-hour composite of the influent

## DARTMOUTH WASTEWATER TREATMENT PLANT RESULTS OF LABORATORY ANALYSES (mg/l)

۰.

	1978	1978		197	79	
	INFLUENT	EFFLUENT		EFFLU	JENT	
PARAMETER	10/24-10/25	10/24-10/25	10/25-10/26	8/27-8/28	8/28-8/29	
COD	510	70	86			
BOD 5	222	13	64	7.2	7.5	
pH (Standard Units)	7.4	7.2	7.2	7.3	7.3	
Suspended Solids	368	9.0	15	1.5	6.0	
Total Solids	762	410	406	374	380	
Settleable Solids (m1/1)		0.0	0.3	0.2	0.2	
Total Alkalinity	192	88	97	57	41	
Total Kjeldahl-Nitrogen	52	12	10	8.9	3.1	
Ammonia-Nitrogen	28	0.77	1	0.33	0.11	
Nitrate-Nítrogen	0.0	15	12	6.6	9.8	
Total Phosphorus	10	12	12	4.3	5.0	
Total Coliform (#/100 ml)		3,000		30	60	
Fecal Coliform (#/100 ml)		<5	and tax	< 5	20	
Total Chlorine Residual					1.3	
Flow (MGD)	0.72	0.72	0.73	0.908	0.940	

7

.

' t

' 2

### FAIRHAVEN WASTEWATER TREATMENT PLANT

.

LOCATION:	Arsene Street Fairhaven, Massachusetts							
RECEIVING WATER:	Acushnet River (the inner New Bedford Harbor)							
CAPACITY:	2.1 MGD							
TREATMENT PROCESS:	Extended aeration Chlorination							
SLUDGE DISPOSAL:	Landfilled at municipal site							
TYPE OF SAMPLE:	Two 24-hour composite samples of the effluent Two grab samples after chlorination for coliform bacteria							
COMMENTS:	No chlorine was found in the effluent							

•

### FAIRHAVEN WASTEWATER TREATMENT PLANT

۰.

μ.

RESULTS OF LABORATORY ANALYSES (mg/1)

	1978	19	78	1979		
	INFLUENT	EFFL	UENT	EFFLUENT		
PARAMETER	10/24-10/25	10/24-10/25	10/25-10/26	8/27-8/28	8/28-8/29	
COD	560	320	320	-	andro solare	
BOD 5	140	20	17	17	11	
pH (Standard Units)	7.1	7.5	7.5	7.1	7.2	
Suspended Solids	138	10	11	14	7.3	
Total Solids	3,128	2,858	3,250	3,164	3,358	
Settleable Solids $(m1/1)$		0.0	0.0	0.1	0.1	
Total Alkalinity	146	133	152	20	73	
Total Kjeldahl-Nitrogen	42	28	16	15	10	
Ammonia-Nitrogen	24	15	14	3.2	3.8	
Nitrate-Nitrogen	0.11	3.1	4.0	20	1.2	
Total Phosphorus	4.1	7.5	5.9	2.3	2.3	
Total Coliform (#/100 ml)		7,500	alla mur	1.5 x 10	<sup>6</sup> <10	
Fecal Coliform (#/100 ml)		<5		0.3 x 10	<sup>6</sup> <5	
Total Chlorine Residual				0.0	Think wrong	
Flow (MGD)	1.13	1.13	0.95	1.07		

### MARION WASTEWATER TREATMENT PLANT

LOCATION:	43 Pumping Station Road Marion, Massachusetts	
RECEIVING WATER:	Tributary to Aucoot Cove	
CAPACITY:	0.34 MGD	
TREATMENT PROCESS:	Stabilization ponds Sand filters Chlorination	
SLUDGE DISPOSAL:	Landfilled at municipal site	
TYPE OF SAMPLE:	Two 24-hour composites of the effluent Two grab samples after chlorination for coliform bacteria	
	one grab sample of the efficient	

10

# MARION WASTEWATER TREATMENT PLANT RESULTS OF LABORATORY ANALYSES (mg/l)

ε.

	1978	1978	1979	
	INFLUENT	EFFLUENT	EFFLUENT	
PARAMETER	10/25	10/25	8/27-8/28	8/28-8/29
COD	230	120		
BOD <sub>5</sub>	120	30	14	9.0
pH (Standard Units)	7.1	6.9	7.6	6.9
Suspended Solids	138	24	12	12
Total Solids	666	362	292	296
Settleable Solids (ml/l)	_~	0.0		10
Total Alkalinity	134	33	41	43
Total Kjeldahl-Nitrogen	34	4.2	5.3	5.6
Ammonia-Nitrogen	23	1.4	1.2	1.5
Nitrate-Nitrogen	0.9	0.4	1.2	1.6
Total Phosphorus	3.9	2.2	2.7	2.7
Total Coliform (#/100 ml)		<b>&lt;10</b>	200	20
Fecal Coliform (#/100 ml)	<b></b>	< 5	< 5	< 5
Total Chlorine Residual			1.5	1.5
Flow (MGD)	0.28	0.28	0.34	0.35

.

۰,

` *\** 

### NEW BEDFORD WASTEWATER TREATMENT PLANT

LOCATION:	Fort Rodman New Bedford, Massachusetts
RECEIVING WATER:	New Bedford Harbor
CAPACITY:	30 MGD
TREATMENT PROCESS:	Primary Chlorination
SLUDGE DISPOSAL:	Landfilled on site
TYPE OF SAMPLE:	Two 24-hour composite samples of the effluent
	Two grab samples after chlorination for coliform bacteria.
	One grab sample of the influent
	One 24-hour composite sample of the effluent

۳.,

# NEW BEDFORD WASTEWATER TREATMENT PLANT RESULTS OF LABORATORY ANALYSES (mg/1)

	1978	1978	1979	
	INFLUENT	EFFLUENT	EFFLUE	INT
PARAMETER	10/24	10/23-10/24	8/27-8/28	8/28-8/29
COD	323	274		
BOD 5	150	<b>9</b> 6	54	87
pH (Standard Units)	7.5	7.1	7.0	6.9
Suspended Solids	92	90	74	90
Total Solids	906	1,846	4,662	2,790
Settleable Solids (ml/l)		0.9	1.9	0.8
Total Alkalinity	121	68	61	97
Total Kjeldahl-Nitrogen	35	32	21	25
Ammonia-Nitrogen	16	13	12	11
Nitrate-Nitrogen	0.0	0.0	0.1	0.1
Total Phosphorus	4.9	2.5	3.9	3.8
Total Coliform (#/100 ml)	unter Stat	430	2.4 x 10 <sup>6</sup>	$0.93 \times 10^{6}$
Fecal Coliform (#/100 ml)		<36	$0.93 \times 10^{6}$	$0.24 \times 10^6$
Flow (MGD)	22	22	24.8	24.5

,

.

.

`,

• 1

### WAREHAM WASTEWATER TREATMENT PLANT

LOCATION:	Route 6 Wareham, Massachusetts
RECEIVING WATER:	Agawam River
CAPACITY:	1.75 MGD
TREATMENT PROCESS:	Extended aeration Sand filters Chlorination
SLUDGE DISPOSAL:	Landfilled at municipal site
TYPE OF SAMPLE:	Two 24-hour composite samples of the effluent Two grab samples after chlorination for bacteria Three 24-hour composite samples of the effluent One 24-hour composite sample of the influent

# WAREHAM WASTEWATER TREATMENT PLANT

۰.

۰,

RESULTS OF LABORATORY ANALYSES (mg/1)

	1978		1978		19	79
	INFLUENT	EFFLUENT			EFFLUENT	
PARAMETER	10/28	10/24-10/25	10/25-10/26	10/26-10/27	8/27-8/28	8/28-8/29
COD		48	5.0			
BOD 5		4.8	4.8		14	1.8
pH (Standard Units)	7.2	4.8	5.0		6.7	5.6
Suspended Solids	130	13	5.0	1.0	23	2.0
Total Solids	560	420	440	412	534	420
Settleable Solids (ml/l)		0.0	0.1	0.0	0.2	0.0
Total Alkalinity	159	2.0	4.0		21	6
Total Kjeldahl-Nitrogen	50	3.6	6:0		6.1	4.5
Ammonia-Nitrogen	31	1.4	2.3		1.1	1.6
Nitrate-Nitrogen	0.0	19	19		19	14
Total Phosphorus	5.1	5.8	0.95		7.0	2.2
Total Coliform (#/100 ml)				<5		<5 <sub>.</sub>
Fecal Coliform (#/100 ml)				<5		<5
Total Chlorine Residual					1.5	1.5
Flow (MGD)	0.26	0.27	0.26	0.26	0.385	0.427

۰.,

• •

## ACUSHNET COMPANY, GOLF DIVISION

LOCATION:	Slocum Street Acushnet, Massachusetts
RECEIVING WATER:	Acushnet Ríver
CAPACITY:	Varies with flow
TREATMENT PROCESS:	pH adjustment
SLUDGE DISPOSAL:	Not applicable
TYPE OF SAMPLE:	A grab sample of the effluent from discharge 010

16

.

# ACUSHNET COMPANY, GOLF DIVISION, ACUSHNET RESULTS OF LABORATORY ANALYSES (mg/l)

.

.

	EFFI	LUENT
PARAMETER	10/26/78	8/28/79
COD		48
pH (Standard Units)		7.7
Suspended Solids		6.5
Total Solids		108
Settleable Solids (ml/l)		0.0
Ammonia-Nitrogen,		0.33
Nitrate-Nitrogen		0.1
Total Phosphorus		0.3
Oil and Grease		10
Lead		0.00
Nickel	<b></b>	0.00
Zinc	0.0	0.02
Flow (MGD)		1.28

. e

# ACUSHNET COMPANY, RUBBER DIVISION

LOCATION:	744 Belleville Avenue New Bedford, Massachusetts
RECEIVING WATER:	Acushnet River
CAPACITY:	Varies with flow
TREATMENT PROCESS:	None
SLUDGE DISPOSAL:	Not applicable
TYPE OF SAMPLE:	A grab sample of the effluent

18

# ACUSHNET COMPANY, RUBBER DIVISION, NEW BEDFORD RESULTS OF LABORATORY ANALYSES (mg/1)

1979

	EFFLUENT
PARAMETER	8/29
COD	5
pH (Standard Units)	7.9
Suspended Solids	6.5
Total Solids	92
Total Alkalinity	23
Total Phosphorus	0.25
Oil and Grease	6.0
Chromium, Total	0.00
Chromium, Hexavalent	0.000
Flow (MGD)	0.718

۰.

. 0

### CHAMBERLAND MANUFACTURING COMPANY

.

LOCATION:	117 King Street New Bedford, Massachusetts
RECEIVING WATER:	Nash Road Pond and Copper Brook
CAPACITY:	Varies with flow
TREATMENT PROCESS:	Oil Separator
TYPE OF SAMPLE:	Grab sample of the effluent from discharge 002

. .

72

.

.

## CHAMBERLAND MANUFACTURING COMPANY RESULTS OF LABORATORY ANALYSES (mg/1)

		EFFLUENT
PARAMETER		10/25/78
Chromium,	Total	0.00
Chromium,	Hexavalent	0.00
Zinc		0.23

.

. ~o

21

-

### LINCOLN PARK AMUSEMENT COMPANY

.

LOCATION:	Route 6 North Dartmouth, Massachusetts
RECEIVING WATER:	East Branch of the Westport River
CAPACITY:	0.05 MGD
TREATMENT PROCESS:	Clarifier Imhoff tank Trickling filter Floculation tank Sand filters Aeration lagoon Chlorination
SLUDGE DISPOSAL:	Landfilled at municipal site.
TYPE OF SAMPLE:	A single grab sample of the effluent A grab sample after chlorination for coliform bacteria

22

## LINCOLN PARK AMUSEMENT COMPANY RESULTS OF LABORATORY ANALYSES (mg/1) 1979

EFFLUENT PARAMETER 8/28 BOD<sub>5</sub> 32 pH (Standard Units) 7.2 Suspended Solids 12 Total Solids 380 Total Kjeldahl-Nitrogen 10 Ammonia-Nitrogen 6.3 Nitrate-Nitrogen 19 Total Phosphorus 2.9 Total Coliform (#/100 ml) 400 Fecal Coliform (#/100 ml) <5 Flow (MGD) 0.05

23

, fin

•

### MASSACHUSETTS MARITIME ACADEMY

LOCATION:	Academy Road, Taylors Point Bourne, Massachusetts
RECEIVING WATER:	Cape Cod Canal
CAPACITY:	0.125 MGD
TREATMENT PROCESS:	Extended aeration Chlorination
SLUDGE DISPOSAL:	Removed by contractor and landfilled at municipal site
TYPE OF SAMPLE:	A single grab sample of the effluent A grab sample after chlorination for coliform bacteria

.

4.

# MASSACHUSETTS MARITIME ACADEMY RESULTS OF LABORATORY ANALYSES (mg/1)

	EFFLU	JENT
PARAMETER	10/26/78	8/27/79
COD	110	
BOD <sub>5</sub>	35	4.5
pH (Standard Units)	7.5	6.3
Suspended Solids	82	21
Total Solids	310	312
Settleable Solids (ml/l)	0.1	0.2
Total Alkalinity	211	10
Total Kjeldahl-Nitrogen	59	2.3
Ammonia-Nitrogen	43	0.04
Nitrate-Nítrogen	3.8	25
Total Phosphorus	5.4	6.0
Total Coliform (#/100 ml)		<36
Fecal Coliform (#/100 ml)		<36
Flow (MGD)		0.125

(<sup>jin</sup>

•

### OLD ROCHESTER HIGH SCHOOL

۲

2.fr

LOCATION:	Marion Road Mattapoisett, Massachusetts
RECEIVING WATER:	Tributary to Mattapoisett Harbor
CAPACITY:	0.012 MGD
TREATMENT PROCESS:	Extended aeration
TYPE OF SAMPLE:	Grab sample of the effluent

•

## OLD ROCHESTER HIGH SCHOOL RESULTS OF LABORATORY ANALYSES (mg/1)

	EFFLUENT
PARAMETER	10/25/78
COD	100
BOD <sub>5</sub>	10
pH (Standard Units)	7.1
Suspended Solids	57
Total Solids	312
Settleable Solids (m1/1)	0.3
Total Alkalinity	49
Total Kjeldahl-Nitrogen	16
Ammonia-Nitrogen	8.8
Nitrate-Nitrogen	5.7
Total Phosphorus	2,2

----

.

 $\mathcal{I}^{\underline{\mu}}$ 

### REVERE COPPER AND BRASS

LOCATION:	24 North Front Street New Bedford, Massachusetts
RECEIVING WATER:	Acushnet River
CAPACITY:	Varies with flow
TREATMENT PROCESS:	Oil separator Metals precipitation
TYPE OF SAMPLE:	Grab samples taken from each of the three discharges, 002, 002A, and 004B

## REVERE COPPER AND BRASS RESULTS OF LABORATORY ANALYSES (mg/1)

	EFFLUENT 10/23/78		E	EFFLUENT		
			8/28/79			
PARAMETER	002	002A	002	002A	004B	
pH (Standard Units)			7.1	1,2	7.0	
Suspended Solids			0.5		11	
Total Solids			386		308	
Settleable Solids (ml/l)					0.05	
Total Phosphorus			0.40		0.30	
Oil and Grease		an		1.8	6.4	
Chromium, Total			0.00	0.00		
Chromium, Hexavalent			0.000	0.000	1000 (1997)	
Copper	0.10	0.12	0.24	0.15		
Zinc	0.05	0.00	0.10	0.01		
Flow (MGD)	50 cm		0.145	0.023	0.032	

29

*,*13

#### GLOSSARY OF TERMS

- <u>Acidity</u> The quantitative capacity of aqueous solutions to react with hydroxyl ions. It is measured by titration with a standard solution of a base to a specified end point. Usually expressed as milligrams per liter of calcium carbonate.
- <u>Alkalinity</u> The capacity of water to neutralize acids, a property imparted by the water's content of carbonates, bicarbonates, hydroxides, and occasionally borates, silicates, and phosphates. It is expressed in milligrams per liter of equivalent calcium carbonate.
- Anaerobic Waste Treatment Waste stabilization brought about through the action of microorganisms in the absence of air or elemental oxygen. Usually refers to waste treatment by methane fermentation.
- <u>Biochemical Oxygen Demand (BOD)</u> The quantity of oxygen used in the biochemical oxidation of organic matter in a specified time, at a specified temperature, and under specified conditions.
- <u>Biological Wastewater Treatment</u> Forms of wastewater treatment in which bacterial or biochemical action is intensified to stabilize, oxidize, and nitrify the unstable organic matter present. Intermittent sand filters, contact beds, trickling filters, and activated sludge processes are examples.
- <u>Chemical Oxygen Demand (COD)</u> A measure of the oxygen-consuming capacity of inorganic and organic matter present in water or wastewater. It is expressed as the amount of oxygen consumed from a chemical oxidant in a specific test. It does not differentiate between stable and unstable organic matter and thus does not necessarily correlate with biochemical oxygen demand.
- <u>Chlorination</u> The application of chlorine to water or wastewater, generally for the purpose of disinfection, but frequently for accomplishing other biological or chemical results.
- <u>Clarification</u> Any process or combination of processes, the primary purpose of which is to reduce the concentration of suspended matter in a liquid.
- <u>Coliform</u> Bacteria found in abundance in the intestinal tract of warmblooded animals. They are not harmful in themselves, but their presence indicates that pathogenic bacteria may be present. Since they can be detected by relatively simple test procedures, coliforms are used to indicate the extent of bacterial pollution from sewage. Bacterial tests usually measure the fecal and total coliforms. Fecal coliform make up about 90 percent of the coliforms discharged in fecal matter. Non-fecal coliforms may originate in soil, grain, or decaying vegetation.
- <u>Comminution</u> The process of cutting and screening solids contained in the wastewater flow before it enters the flow pumps or other units in the treatment plant.

- <u>Composite Wastewater Sample</u> A combination of individual samples of water or wastewater taken at selected intervals, generally hourly, for some specified period, to minimize the effect of the variability of the individual sample. Individual samples may have equal volume or be proportioned to the flow at the time of sampling.
- Data Records of observations and measurements of physical facts, occurrences, and conditions, reduced to written, graphical, or tabular form.
- Fats (wastes) Triglyceride esters of fatty acids; erroneously used as synonomous with grease.
- <u>Flocculation</u> In water and wastewater treatment, the agglomeration of colloidal and finely divided suspended matter after coagulation by gentle stirring by either mechanical or hydraulic means. In biological wastewater treatment where coagulation is not used, agglomeration may be accomplished biologically.
- <u>Grab Sample</u> A single sample of wastewater taken at neither set time nor flow.
- <u>Grease</u> In wastewater, a group of substances including fats, waxes, free fatty acids, calcium and magnesium scaps, mineral oils, and certain other nonfatty materials. The type of solvent and method used for extraction should be stated for quantification.
- <u>Grit Chamber</u> A detention chamber or enlargement of a sewer designed to reduce the velocity of flow of the liquid to permit the separation of mineral from organic solids by differential sedimentation.
- <u>Hardness</u> A characteristic of water imparted by salts of calcium, magnesium, and iron such as bicarbonates, carbonates, sulfates, chlorides, and nitrates, that cause curdling of soap, deposition of scale in boilers, damage in some industrial processes, and sometimes objectionable taste. It is expressed as equivalent calcium carbonate.
- <u>Heavy Metals</u> These elements are toxic when present in sufficient quantities and can be fatal. They can adversely affect sewage treatment systems and the biological systems of waterbodies. They include cadmium, chromium, copper, iron, lead, manganese, nickel, and zinc.
- <u>Industrial Wastes</u> The liquid wastes from industrial processes, as distinct from domestic or sanitary wastes.
- <u>Inorganic Matter</u> Chemical substances of mineral origin, or, more correctly, not of basically carbon structure.
- Lagoon A pond containing raw or partially treated wastewater in which aerobic or anaerobic stabilization occurs.
- <u>Most Probable Number (MPN)</u> That number of organisms per unit volume that, in accordance with statistical theory, would be more likely than any other number to yield the observed test result with the greatest frequency. Expressed as density of organisms per 100 ml. Results are computed from the number of positive findings of coliform-group organisms resulting from multiple-portion decimal-dilution plantings.

- <u>Nitrogen</u> A common non-metallic element that in free form is normally a colorless, odorless, tasteless, insoluble, inert, diatomic gas. In the combined form, it has a wide range of valences and is a constituent of biologically important compounds (as proteins) and hence of all living cells as well as industrially important substances (as cyanides, fertilizers, dyes).
- <u>Nitrogen, Ammonia</u> A compound of nitrogen and hydrogen, NH<sub>3</sub>, which is part of the nitrogen cycle. Its presence in sufficient amounts in a stream can indicate a wastewater discharge. The oxidation of ammonia depletes a stream of dissolved oxygen. It is toxic in sufficient amounts, especially to fish.
- <u>Nitrogen, Kjeldahl</u> This represents the total organic nitrogen content of water.
- Nitrogen, Nitrate Nitrate represents the most highly oxidized phase in the nitrogen cycle and normally reaches important concentrations in the final stages of biological oxidation. Nitrogen in this form is readily available to plants.
- Organic Matter Chemical substances of animal or vegetable origin, or more correctly, of basically carbon structure, comprising compounds consisting of hydrocarbons and their derivatives.
- Oxidation The addition of oxygen to a compound. More generally, any reaction which involves the loss of electrons from an atom.
- Oxidation Pond A basin used for the retention of wastewater before final disposal, in which biological oxidation of organic matter is affected by natural or artificially accelerated transfer of oxygen to the water from air.
- Parshall Flume A calibrated device developed by Parshall for measuring the flow of a liquid in an open conduit.
- Pathogenic Bacteria Bacteria that may cause disease in the host organism by their parasitic growth.
- <u>pH</u> The reciprocal of the logarithm of the hydrogen ion concentration. The concentration is the weight of hydrogen ions in grams per liter of solution. Neutral water, for example, has a pH value of 7 and hydrogen ion concentration of 10<sup>-7</sup>.
- <u>Phenol</u> An aromatic compound which is a monohydroxy derivative of benzene. In concentrated solution, it is quite toxic to bacteria. Widely used as a germicide. Commonly known as carbolic acid.
- <u>Phosphorus</u> A nonmetallic multivalent element of the nitrogen family that occurs widely in combined form, especially as inorganic phosphates in minerals, soils, and natural waters, and as organic phosphates in all living cells; it exists in several allotropic forms. The majority of

Se 6

the phosphorus contained in domestic sewage and industrial wastes comes from detergents.

- Primary Settling Tank The first settling tank for the removal of settleable solids through which wastewater is passed in a treatment works.
- <u>Primary Treatment</u> The first major (sometimes the only) treatment in a wastewater treatment works, usually sedimentation. The removal of a substantial amount of suspended matter but little or no colloidal and dissolved matter.
- <u>Residual Chlorine</u> Chlorine remaining in water or wastewater at the end of a specified contact time as combined or free chlorine.
- Sampler A device used with or without flow measurement to obtain an aliquot portion of water or waste for analytical purposes. May be designed for taking a single sample (grab), composite sample, continuous sample, or periodic sample.
- <u>Secondary Settling Tank</u> A tank through which effluent from some prior treatment process flows for the purpose of removing settleable solids.
- Secondary Wastewater Treatment The treatment of wastewater by biological methods after primary treatment by sedimentation.
- <u>Sludge Digestion</u> The process by which organic or volatile matter in sludge is gasified, liquified, mineralized, or converted into more stable organic matter through the activities of either anaerobic or aerobic organisms.
- <u>Sludge Thickening</u> The increase in solids concentration of sludge in a sedimentation or digestion tank.
- Solids, Settleable That matter in wastewater which will not stay in suspension during a pre-selected settling period, such as an hour, but which either settles to the bottom or to the top. In the Imhoff cone test, the volume of matter that settles to the bottom in one hour.
- Solids, Suspended Solids that either float on the surface of, or are in suspension in, water, wastewater, or other liquids and which are largely removable by laboratory filtering. The quantity of material removed from wastewater in a laboratory test, as prescribed in <u>Standard Methods for the Examination of Water and Wastewater</u>, and referred to as non-filterable residue.
- Solids, Total The sum of dissolved and undissolved constitutents in water or wastewater, usually stated in milligrams per liter.
- <u>Wastewater Survey</u> An investigation of the quality and characteristics of each waste stream, as in an industrial plant or municipality.